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## An easy and versatile approach to the synthesis of chiral pheromone lactones via 4,4-dimethyl-2-oxazoline derivatives

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Abstract—As part of our ongoing investigation on the versatility of 4,4-dimethyl-2-oxazoline derivatives, we present a straightforward synthesis of chiral lactone pheromones from readily available starting materials. As an application, we describe the diastereoselective synthesis of cis and trans-2-methyl-5-hexanolide  $(1)$ , a pheromone component of the carpenter bee Xylocopa hirutissima, and a formal synthesis of (4R,5Z)-5-tetradecen-4-olide (2), the sex pheromone of the Japanese beetle Popillia japonica. 2004 Elsevier Ltd. All rights reserved.

Lactone structures are found in many natural products among which there are a number with strong biological activity.[1](#page-1-0) Lactones were isolated from insects and are part of the intriguing communication systems of bees, beetles and other taxa.<sup>[2](#page-1-0)</sup> During previous years, several methodologies to synthesize lactones were developed.[3](#page-1-0) Here we describe a way to synthesize chiral lactones, employing a coupling reaction of 4,4-dimethyl-2-oxazoline derivatives with alkyl iodides as the key step (Scheme 1). We choose cis- and trans-2-methyl-5-hexanolide (1) as an example of a diastereoselective synthesis, and a formal synthesis of (4R,5Z)-5-tetradecen-4 olide (2) as an enantioselective one.

While compound *cis*-2-methyl-5-hexanolide (1) is a component of the pheromone blend of the carpenter bee  $Xylocopa hirutissima<sup>4</sup> (4R,5Z)-5-tetradecen-4-olide (2)$  $Xylocopa hirutissima<sup>4</sup> (4R,5Z)-5-tetradecen-4-olide (2)$  $Xylocopa hirutissima<sup>4</sup> (4R,5Z)-5-tetradecen-4-olide (2)$ 



Scheme 1. General strategy to chiral lactones.

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is the sex pheromone of the Japanese beetle Popillia japonica. [5](#page-1-0)

All four possible isomers of (1) were synthesized using the optically pure hydroxy iodides  $(R)$ - and  $(S)$ - $(4)$ , obtained from PHB and by microbiological reduction of ethyl acetoacetate, respectively,  $6 \text{ as}$  $6 \text{ as}$  electrophilic source. The reaction of 2-ethyl-4,4-dimethyl-2-oxazoline



Scheme 2. Synthesis of the lactones  $(2R,5S)-(1a)/(2S,5S)-(1b)$  and  $(2S,5R)-(1c)/(2R,5R)-(1d).$ 

Keywords: Pheromones; Synthesis; Chiral lactones; Oxazoline.

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Scheme 3. Formal synthesis of lactone  $(2)$ .

(3) with  $(S)$ -(4) afforded compound (5) in 92% isolated yield, $\frac{7}{7}$  which was hydrolyzed and cyclized in acidic media ('one-pot'), to furnish compound (1) in 85% yield as a mixture of stereoisomers (1a) and (1b). Similarly, isomers (1c) and (1d) were synthesized in 87% yield using iodide  $(R)$ - $(4)$  ([Scheme 2\)](#page-0-0). Mixtures of diastereoisomers were obtained in a ratio of 70:30 (1a/b or 1c/d) as determined by GC and  ${}^{1}H$  NMR analysis.<sup>8</sup>

Optimal yields (>90%) were obtained when compound (5) was not isolated, but rather hydrolyzed and cyclized straight from the alkylation step. Methods for the separation of diastereoisomers are largely described in the literature.<sup>9</sup>

A slightly different approach was used to synthesize  $(4R, 5Z)$ -5-tetradecen-4-olide (2). In this case,  $(R)$ -4hydroxymethyl-2,2-dimethyl-1,3-dioxolane (6) was employed as a chiral element. Iodide (7) was obtained from commercial  $(6)$  in 65% yield.<sup>10</sup> The lithium anion was generated from 2,4,4-trimethyl-2-oxazoline (3a) and alkylated in situ with iodide (7), yielding (8) in 85%.<sup>11</sup> 'One-pot' hydrolysis followed by cyclization under acidic conditions provided the known hydroxy lactone (9)  $([\alpha]_D$  -31.0 (c 0.4, EtOH); lit.<sup>12</sup>  $[\alpha]_D$  $-33 \pm 2$  (c 2.9, EtOH)) in 60% yield,<sup>13</sup> which can be employed in the syntheses of chiral pheromone lactones like  $(2)$  and others<sup>14,15</sup> (Scheme 3).

In conclusion, we have explored the versatility of oxazoline derivatives in the synthesis of five and six-membered ring lactones. The examples given here demonstrate a useful application of this methodology.

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- 7. Selected spectroscopic data for compound  $(5)$ . <sup>1</sup>H NMR  $(400 \text{ MHz}, \text{CDC1}_3)$ :  $\delta$  1.2 (d, J = 6.8 Hz, 6H), 1.3 (s, 3H), 1.4–1.9 (m, 4H), 2.37–2.5 (m, 1H), 3.72–3.82 (m, 1H), 3.9 (s, 2H);  $^{13}$ C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  17.74, 18.18, 23.33, 23.53, 28.12, 28.29, 29.84, 29.89, 33.28, 33.43, 36.50, 36.93, 66.56, 66.95, 67.33, 78.80, 78.85, 169.68. IR ( $v_{\text{max}}$ ) film, cm<sup>-1</sup>) 3469, 2969, 2930, 1659. MS m/z (%) 200  $(M^+ + 1, 17)$ , 184 (12), 166 (21), 140 (48), 127 (100), 112 (32), 98 (21).
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